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Principles of Chemistry: A Molecular Approach

Principles of Chemistry: A Molecular Approach (Tro) Chapter 2 Atoms and Elements 1) Which of the following is an example of the law of multiple proportions? A) A sample of chlorine is found to contain three times as much Cl-35 as Cl-37 B) Two different compounds formed from carbon and oxygen have the following mass ratios:

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Computational chemistry principles extended to molecular simulation are not included in this book; we hope that a sister publication of this book covering that aspect will be released in the near future We have tried to make the explanations clear and complete to the satisfaction of the reader However, re-

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guiding principles of chemistry as experienced in all parts of our everyday life B To train them in the basic aspects of chemical structure and to enable them to Define and distinguish between atomic mass and molecular mass, gram-atom and gram-mole 2 Determine the mass of reactants required for a reaction 3 Determine the precise amount

AP CHEMISTRY 2008 SCORING GUIDELINES

2 is a molecular solid with only weak dipole-dipole and London forces among SO₂ molecules One point is earned for recognizing that SiO₂ is a covalent network solid, and that strong covalent bonds must be broken for SiO₂ to melt

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BASIC PRINCIPLES OF ORGANIC CHEMISTRY

of the principles of thermodynamics, quantum mechanics, kinetics, and spectroscopy to understanding and correlating the myriad of seemingly unrelated facts of organic chemistry Much of this material could be appropriately categorized as belonging to a "Department of Fuller Explanation," and rightly so because it rep-

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AP Chemistry Intermolecular Forces FRQs

principles of chemical bonding and/or intermolecular forces In each part, your answer must include references to both substances (a) (b) (c) (d) Even though NH₃ and CH₄ have similar molecular masses, NH₃ has a much higher normal boiling point (-33°C than (-164°C At 25°C and 10 atm, ethane (C₂H₆) is a gas and hexane (C₆H₁₄) is a liquid

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Clinical Chemistry

clinical chemistry, which encompasses a wide variety of tests and is a major area of concentration in hospital and reference core laboratories Clinical chemistry uses many different methodologies, manual and fully automated tests, examines both very common and esoteric analytes, mixes basic chemistry with biochemistry, engineering, informatics and

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